Loma Linda University **2011 Water Quality Report**

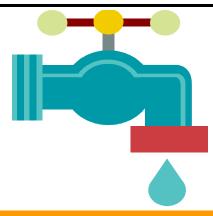
We are proud to present our annual water quality report covering all testing performed between January 1 and December 31, 2011. We are pleased that the safe and reliable water supplied to Loma Linda University, Loma Linda University Medical Center, and many other related entities on the Loma Linda University campus has met or exceeded all State and Federal drinking water health standards. We remain vigilant in safeguarding our water supply.

THE WATER SYSTEM

The Loma Linda University water system consists of two wells, a 1.4 million gallon storage tank, and many miles of pipeline that service approximately 300 residents in student housing year-round and a substantial transient population. The Loma Linda University wells are located in the Bunker Hill Basin. The Bunker Hill Basin is a natural underground aquifer that is replenished from annual rainfall and snow pack from the San Bernardino Mountain range. Loma Linda University also uses a supplementary supply of water from the City of Loma Linda when necessary.

SPECIAL HEALTH INFORMATION: Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. USEPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791.

EN ESPAÑOL: Este informe contiene información muy importante sobre su aqua potable. Tradúzcalo o hable con alguien que lo entienda bien.



PPM & PPB?

Parts per million (ppm) and parts per billion (ppb) are units of concentration used in measuring levels of contaminants in water. One ppm or milligram per liter (mg/L) is one part per 1,000,000. One ppb or microgram per liter (ug/L) is one part per 1,000,000,000.

To gain a better understanding of the scale involved with ppm and ppb, here are a few analogies.

One ppm is like:

- 1 drop in 17 gallons of water
- 1 inch in 16 miles
- 1 second in 11.5 days
- 1 car in bumper to bumper traffic from San Francisco to Cleveland
- 1 penny in \$10,000

One ppb is like:

- 1 drop in 17,000 gallons of water
- 1 second in 32 years
- 1 penny in \$10,000,000

l ppm = 1 mg/L = 1 part per 1,000,000 l ppb = 1 mg/L = 2

Water Quality

In order to ensure that tap water is safe to drink, the U.S. Environmental Protection Agency (USEPA) and the state Department of Public Health (Department) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. Depart-

ment regulations also establish limits for contaminants in bottled water that must provide the same protection for public health.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity. Contaminants that may be present in source water include:

Microbial Contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.

Inorganic contaminants, such as salts and metals, that can be naturally occurring or result from urban storm water runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.

Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff, and residential uses.

Organic chemical contaminants, including synthetic and volatile organic chemicals, that are by-products of industrial processes and petroleum production and can also come from gas stations, urban storm-water runoff, agricultural application, and septic systems.

Radioactive contaminants, which can be naturally occurring or the result of oil and gas production and mining activities.

Abbreviations and Definitions

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (USEPA).

MRDL (Maximum Residual Disinfectant Level): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

MRDLG (Maximum Residual Disinfectant Level Goal): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

ND: not detectable at testing limit

Notification Level: The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

pCi/L: picocuries per liter (a measure of radiation)

PHG (Public Health Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

ppm: parts per million or milligrams per liter (mg/L).

ppb: parts per billion or micrograms per liter (ug/L).

Primary Drinking Water Standards (PDWS): MCLs or MRDL's for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

Secondary Drinking Water Standards (SDWS): MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

SAFE DRINKING WATER HOTLINE:

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline at (800) 426-4791.

2011 DRINKING WATER QUALITY TEST RESULTS

In 2011, the Loma Linda University Power Plant conducted many water quality tests from samples taken at various locations throughout the water system in accordance with state and federal regulations. We are pleased that our water complied with or did better than those regulations. The following chart shows contaminants that were detected in the water. The State allows us to monitor for some contaminants less than once per year because concentrations of these contaminants do not change frequently. Some of our data, though representative, are more than one year old. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk.

indicate that the water poses						
	REGU	LATED CON	ITAMINAN	TS WITH PRI	MARY MC	L'S
Radioactive Contaminants						
	LLU Average	LLU Range	MCL	PHG (MCLG)	Violation	Typical Source of Contaminant
Gross Alpha (pCi/L) (2008)	3.7	2.06 - 5.24	15	N/A	No	Erosion of natural deposits
norganic Contaminants						
	LLU Average	LLU Range	MCL	PHG (MCLG)	Violation	Typical Source of Contaminant
Total Chromium (ppb) (2011)	2.9	1.6 - 5.2	50	N/A	No	Discharge from steel and pulp mills and chrome plating; erosion of natural deposits
Fluoride (ppm) (2011)	0.77	0.6 - 0.9	2	1	No	Erosion of natural deposits; water additive which promotes strong teeth; discharge from fertilizer and aluminum factories
Nitrate (ppm) (2011)	20.21	9.5 - 37	45	45	No	Runoff and leaching from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits
Perchlorate (ppb) (2011)	0.49	ND - 4.4	6	6	No	Inorganic chemical used in solid rocket propellant, fireworks, explosives, flares, matches, and a variety of industries. It usually gets into drinking water as a result of environmental contamination from historic aerospace or other industrial operations that used or use, store, or dispose of perchlorate and its salts
Disinfection Byproducts, Dis	sinfectant Resi	duals, & Disir	nfection By	product Precu	ırsors	, , , , , , , , , , , , , , , , , , , ,
,	LLU Average	LLU Range	MRDL	MRDLG	Violation	Typical Source of Contaminant
Total Trihalomethanes (TTHMs) (ppb) (2011)	1.1	1.1	80	N/A	No	By-product of drinking water disinfection
	REGULA	ATED CONT	AMINANTS	S WITH SECC	NDARY M	ICL'S
	LLU Average	LLU Range	Secon	dary MCL	Violation	Typical Source of Contaminant
Chloride (ppm) (2011)	38	35 - 42	500		No	Runoff/leaching from natural deposits
Iron (ppb) (2008)	220	220	300		No	Leaching from natural deposits; industrial waste
Specific Conductance (micromhos) (2011)	370	520 - 770	1,600		No	Substances that form ions when in water
Sulfate (ppm) (2011)	55	44 - 69	500		No	Runoff/leaching from natural deposits; industrial wastes
Total Dissolved Solids (ppm) (2011)	348	200 - 460	1,000		No	Runoff/leaching from natural deposits

Nitrate

Nitrate in drinking water at levels above 45 ppm (mg/L) is a health risk for infants of less than six months of age. Such nitrate levels in drinking water can interfere with the capacity of the infant's blood to carry oxygen, resulting in a serious illness; symptoms include shortness of breath and blueness of the skin. Nitrate levels above 45 ppm may also affect the ability of the blood to carry oxygen in other individuals, such as pregnant women and those with certain specific enzyme deficiencies. If you are caring for an infant, or you are pregnant, you should ask advice from your health care provider.

	SAMPLIN	G RESULTS S	HOWING TH	E DET	ΓEC	TION OF C	COPPER						
	No. of	90th %ile	No. Sites	AL	AL	PHG	Typical Source of Bacteria						
	Samples	Level	Exceeding										
	Collected	Detected	AL										
Copper (ppm) (2010)	5	0.10	0	1.3	3	0.17	Internal corrosion of plumbing systems;						
							erosion of natural deposits; leaching from						
					wood preservatives								
SAMPLING RESULTS FOR SODIUM AND HARDNESS													
Chemical or Constituent	No. of	LLU Average	LLU Range	MCL	CL	PHG	Typical Source of Bacteria						
	Samples					(MCLG)							
0 1	Collected	00	04 00			.							
Sodium (ppm) (2011)	5	83	81—82	None			Refers to the salt present in the water and						
Llordnood () (0044)	5		68—210				s generally naturally occurring.						
Hardness (ppm) (2011)	5	128	00-210	None		(the sum of polyvalent cations present gen-						
							erally magnesium and calcium. The cations						
							are usually naturally occurring.						
UNREGULATED CONTAMINANTS AND OTHER SUBSTANCES													
		LLU Average	LLU Range	9	Notification Leve								
Alkalinity (ppm) (2011)		167	140 - 220)	None		Unregulated contaminant monitoring helps the EPA and the California						
Biocarbonate (ppm) (2011)		203	170 - 260)		None	Department of Public Health to deter-						
Calcium (ppm) (2011)		39	22 - 62		None		mine where certain contaminants occur and whether the contaminants						
Chromium VI (ppb) (2004)		4.25	4.2 - 4.3		None		need to be regulated.						
Magnesium (ppm) (2011)		7.38	3.5 - 13		None								
DH (2011)		8	7.8 - 8.1		None								
Potassium (ppm) (2008)		2	1.6 - 2.8		None								
Radium 228 (pCi/L) (2008)		0.18	0 - 0.532		None								
Bromoform (ppb) (2011)		0.85	0.85		None								

DRINKING WATER SOURCE ASSESSMENT: A drinking water source assessment of both LLU wells was conducted by San Bernardino County, Environmental Health Services in May, 2002. The purpose of this assessment was to identify potential sources of contamination and develop ways to protect the water supply. Our water source is considered most vulnerable to contamination by activities such as sewer collection systems and automobile gas stations. It is also vulnerable from a known contaminant plume that contains perchlorate.

A copy of the complete assessment may be viewed at the San Bernardino County Environmental Health Services office at 385 N. Arrowhead Avenue, 2nd Floor, San Bernardino, CA 92415-0160. You may request a summary of the assessment be sent to you by contacting the Environmental Health Specialist at (909) 387-4666.

LEAD IN WATER

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and plumbing. The Loma Linda University Water System is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. We do conduct random periodic lead testing around the Loma Linda University campus and have never exceeded the MCL.

When water has been sitting for several hours, you can minimize the potential for lead exposure by flushing the tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water at home, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.